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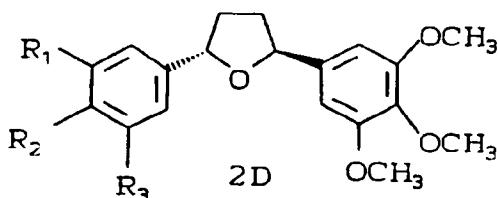
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(54) Process of making 2,5-diaryl tetrahydrofurans and analogs thereof useful as paf antagonists.

(57) The present invention is directed to a process of making tetrahydrofuran of the formula (2D)



resulting in 2,5-diaryl tetrahydrofuran PAF antagonists of pharmaceutically acceptable purity.

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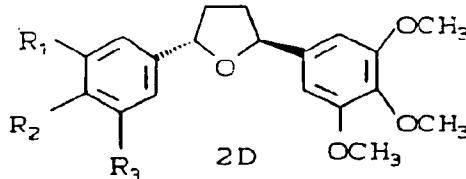
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BACKGROUND OF THE INVENTION

Compounds of formula (2D) are known and are known to possess utility as platelet-activating factor (PAF) antagonists

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Platelet-activating factor (PAF) has recently been identified as an acetyl glyceryl ether phosphorylcholine (AGEPC), i.e., 1-O-hexadecyl/octadecyl-2-acetyl-sn-glyceryl-3-phosphocholine (Hanahan D.J., et al., *J. Biol. Chem.* 255:5514, 1980). PAF has been linked to various biological activities and pathways making it one of

the important mediators responsible for a variety of physiological processes including activation or coagulation of platelets, pathogenesis of immune complex deposition, smooth muscle contraction, inflammation, hypotension, shock, pain, edema as well as respiratory, cardiovascular and intravascular alterations. These physiological

20 processes are in turn associated with a large group of diseases, for example, inflammatory disease, cardiovascular disorder, hypotension, shock, psoriasis, allergic and skin diseases, asthma, lung edema, peptic or

stomach ulcer, dental pain, and adult respiratory distress syndrome.

Some compounds of formula (2D) as well as their utility as PAF antagonists and their method of preparation are disclosed in U.S. Patent 4,539,335 which issued on September 3, 1985; E.P. 0 199 324, which published

25 on October 29, 1986; E.P. 0 322 033, published on June 28, 1989; copending U.S. Application 362,919, filed

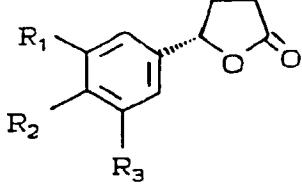
June 8, 1989 and 505,712 filed April 11, 1990 and which issued as on December 11, 1990 as U.S. 4,977,146

all of which are hereby incorporated by reference.

In one aspect the invention concerns the making of Compounds of the following formula

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Corey et al. describes a synthesis of chiral *trans*-2,5-diaryltetrahydrofurans in which a partially optically enriched lactone of the structure above is an intermediate. These chiral 5-aryl- $\gamma$ -butyrolactones are prepared via reduction of 3-(3,4-dimethoxybenzoyl) propionate with a chiral oxaborolidine, then base catalyzed lactonization. See E.J. Corey, R.K. Bakshi, Shibata, C.P. Chen and V.K. Singh; *J. Am. Chem. Soc.* 1987, 109, 7925. This art yields 5-aryl- $\gamma$ -butyrolactones of only 95% optical purity. Further, no method is disclosed for increasing the optical purity to levels of up to and beyond 99.5%. No method for enrichment of optical purity at a later stage in the synthesis of the target 2,5-diaryl- tetrahydrofurans is disclosed.

H.C. Brown et al. describes the use of  $\beta$ -chloro-diisopinocampheylborane to reduce ketones to alcohols at optical purity of 92 to 98%. This methodology, however, was not extended to  $\gamma$ -keto esters. Moreover, this art is ineffective when applied to  $\beta$ -ketoesters. See H.C. Brown, J. Chandrasekharan, and P.V. Ramachandran *J. Am. Chem. Soc.* 1988, 110, 1543 and U.S. 4,772,752 filed August 29, 1986.

M.M. Pompipom et al. discloses the preparation of PAF antagonists of the diaryl tetrahydrofuran structure via an acid catalyzed cyclization of a 1,4-diaryl-1,4-butanediol. This method however produces an isomeric mixture of *cis* and *trans* 2,5-diaryltetrahydrofurans. The optically pure 5-aryl- $\gamma$ -butyrolactones in the present invention are readily converted stereo-selectively to the single, physiologically active *trans* isomer exclusively (M.M. Pompipom, R.L. Bugianesi, J.C. Chabala, *Tetrahedron Letters*, 1988, 29, 6211).

H. Stetter et al. discloses a cyanide catalyzed addition of benzaldehyde derivatives to  $\alpha,\beta$ -unsaturated ketones, nitriles and esters (H. Stetter, M. Schreckenberg, and K. Wiemann, *Chem. Ber.* 109, 541, 1976 and H. Stetter and M. Schreckenberg, *Agnew. Chem.* 85, 89 (1973); *Agnew. Chem. Int. Ed. Eng.* 12, 81, 1973). Application to electron-rich aromatic rings have not been reported. In addition the reported yields are low, ranging

from 34 to 68% (with p-chlorobenzaldehyde).

The present invention provides a method for the preparation of 5-aryl- $\gamma$ -butyrolactones having an optical purity of up to 99.5% enantiomeric excess (ee) and beyond. Yields for the process are high and only commercially available starting materials materials are employed. Both enantiomers can be prepared with equal facility.

5 In a second aspect the invention concerns making a compounds of formula (2D) from the lactone. Preparation of compounds of formula (2D) has been achieved via an acid catalyzed cyclization of a 1,4-diaryl-1,4-butanediol. This method, however, produces an isomeric mixture of *cis* and *trans* 2,5-diaryl tetrahydrofurans. See *Tetrahedron Letters*, 29, p. 6211 (1988). In contrast, the optically pure 5-aryl-butyro-lactones of the present invention are readily converted in a stereoselective reaction to a single physiologically active *trans*-isomer.

10 Some 2,5-diaryltetrahydrofurans have been prepared from 5-aryl substituted furanosides via a direct lactol activation with TMS-Br followed by coupling with an organometallic partner. See *J.Am.Chem.Soc.*, 109, p. 7925 (1987). However, this method, when applied to electron-rich aromatic ring systems such as 3,4,5-trimethoxyphenol, produces low yields and is accompanied by significant side products.

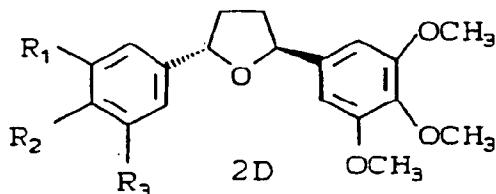
15 Preparation of a furanosyl bromide from a furanosyl acetate with trimethylsilyl bromide has also been described. See *Tetrahedron Letters*, p. 513 (1981). Unfortunately, it is our experience that this technique leads to incomplete bromination with a consequent loss of yield and purity in the subsequent coupling reaction.

In contrast to the prior art, the instant process provides an efficient means to produce optically pure *trans*-2,5-diaryltetrahydrofurans from optically pure 5-aryl- $\gamma$ -butyrolactones in high yield with high stereoselectivity (*trans:cis* greater than 75:1) and high degrees of regiocontrol with respect to substitution on the aromatic ring.

20 Accordingly, the present invention is directed to the preparation of the most potent isomers of known or novel tetrahydrofuran derivatives as PAF antagonists and use them for the treatment of various diseases including prevention of platelet aggregation, hypotension, inflammation, asthma, lung edema, adult respiratory distress syndrome, various shock syndromes, cardiovascular disorders and other related skeletal-muscular disorders, graft-host rejection, nephritis, pancreatitis and lupus.

#### BRIEF DESCRIPTION OF THE INVENTION

The present invention is directed to a process for making to optically pure *trans*-2,5-diaryltetrahydrofuran, Compound 2D



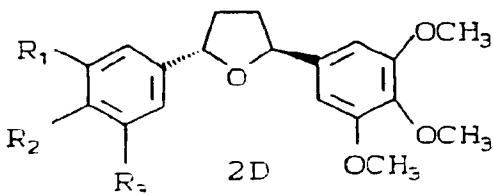
#### BRIEF DESCRIPTION OF THE INVENTION

In one aspect, the invention concerns a process wherein 5-Aryl- $\gamma$ -butyrolactones is stereospecifically converted to *trans*-2,5-diaryl-tetrahydrofurans, which are potent PAF antagonists. In the process, the stereochemistry at C-5 is directly translated to the incipient C-2 stereocenter. The present invention, provides an efficient method of producing 5-aryl- $\gamma$ -butyrolactones of high optical purity which is critical to the synthesis of optically pure PAF antagonists.

In a second aspect the process involves a four step sequence in which the 5-aryl- $\gamma$ -butyro-lactone is activated through its conversion to a glycosyl bromide. The activated bromide is coupled in a stereoselective manner with an aryl metal species to establish the *trans*-2,5 substitution pattern about the tetrahydrofuran ring.

#### DETAILED DESCRIPTION OF THE INVENTION

The present invention is directed to a process of making 2,5-diaryltetrahydrofuran PAF antagonists of pharmaceutically acceptable purity of the following structural formula:



10 wherein:

R<sup>1</sup> is iodide, or bromide, or

R<sup>1</sup> is S(O)<sub>n</sub>R<sub>a</sub> in which n is 0, 1 or 2 and R<sub>a</sub> is selected from the group consisting of

- (a) C<sub>1-6</sub>alkyl,
- (b) C<sub>2-6</sub>alkenyl,
- (c) C<sub>2-6</sub>alkynyl,
- (d) substituted C<sub>1-6</sub>alkyl wherein the substituent is selected from the group consisting of hydroxy, protected hydroxy, N-Cl-4alkylamino, and N,N-Cl-4di-alkylamino,
- (e) C<sub>1-6</sub>alkoxy-C<sub>1-6</sub>alkyl,
- (f) C<sub>1-6</sub>alkylcarbonyl-C<sub>1-6</sub>alkyl; and

20 R<sub>2</sub> is selected from the group consisting of

- (a) C<sub>1-12</sub>alkoxy,
- (b) C<sub>2-6</sub>alkenoxy,
- (c) C<sub>2-6</sub>alkynoxy,
- (d) C<sub>2-6</sub>(halo)<sub>x</sub> alkoxy wherein x is 1,2,3,4 or 5 and halo is chloro fluoro or bromo,
- (e) substituted C<sub>1-6</sub>alkoxy wherein the substituent is hydroxy, or protected hydroxy,
- (f) C<sub>1-6</sub>alkoxy-C<sub>1-6</sub>alkoxy,
- (g) C<sub>1-6</sub>alkyl S(O)<sub>m</sub>-C<sub>1-6</sub>alkoxy in which m is 0, 1 or 2,
- (h) C<sub>1-6</sub>alkyloxysulfonyl-C<sub>1-6</sub>alkoxy,
- (i) C<sub>1-6</sub>alkyl carbonyl-C<sub>1-6</sub>alkoxy,
- (j) phenyl-C<sub>1-6</sub>alkoxy,
- (k) azido-C<sub>1-6</sub>alkoxy,
- (l) cyano-C<sub>1-6</sub>alkoxy,
- (m) C<sub>1-6</sub>alkylS(O)<sub>m</sub>-C<sub>1-6</sub>alkoxy,
- (n) N-substituted, or N,N-disubstituted amino-C<sub>1-6</sub>alkyloxy wherein the substituents are each individually C<sub>1-6</sub>alkyl;

35 R<sub>3</sub> is selected from the group consisting of

- (a) C<sub>1-6</sub>alkoxy,
- (b) substituted C<sub>1-6</sub>alkoxy wherein the substituent is selected from the group consisting of hydroxy, protected hydroxy, N-C<sub>1-4</sub>alkylamino, N,N-diC<sub>1-4</sub>alkylamino,
- (c) -O-C<sub>1-6</sub>alkyl-0-R<sup>10</sup>, wherein R<sup>10</sup> is
  - (1) -PO<sub>2</sub>(OH)<sup>-</sup>M<sup>+</sup> wherein M<sup>+</sup> is a pharmaceutically acceptable cation,
  - (2) -C(O)(CH<sub>2</sub>)<sub>2</sub>-CO<sub>2</sub><sup>-</sup>M<sup>+</sup>, or
  - (3) -SO<sub>3</sub><sup>-</sup>M<sup>+</sup>,
- (d) C<sub>1-6</sub>alkylcarbonyl-C<sub>1-6</sub>alkoxy,
- (e) C<sub>1-6</sub>alkoxyaminocarbonyloxy,
- (f) halophenylC<sub>1-6</sub>alkoxy, and
- (g) C<sub>1-6</sub>carboxyalkoxy.

40 As will be understood by those skilled in the art, pharmaceutically acceptable salts include, but are not limited to salts with inorganic acids such as hydrochloride, sulfate, phosphate, diphosphate, hydrobromide, and nitrate or salts with an organic acid such as malate, maleate, fumarate, tartrate, succinate, citrate, acetate, lactate, methanesulfonate, p-toluenesulfonate or palmoate, salicylate and stearate. Similarly pharmaceutically acceptable cations include, but are not limited to hydrogen, sodium, potassium, calcium, aluminum, magnesium, barium, zinc, lithium, ammonium, or an amino acid including glycine, alanine, valine, leucine, isoleucine, serine, threonine, aspartic acid asparagine, glutamic acid, glutamine lysine, hydroxylysine, histidine,

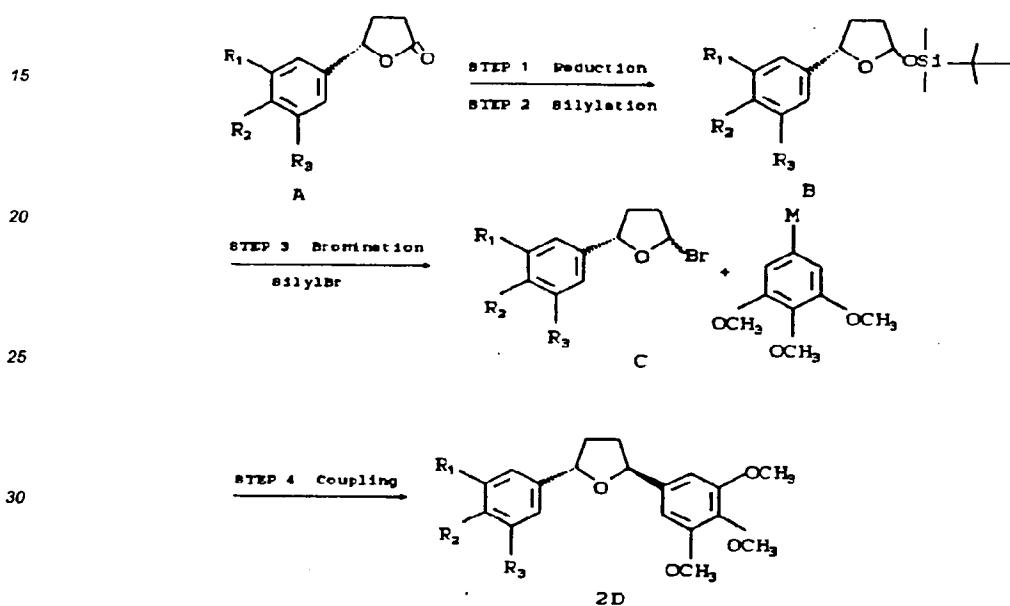
arginine, phenylalanine, tyrosine, tryptophan, thyroxine, cystine, cysteine, methionine, proline, hydroxyproline, ornithine,  $\beta$ -alanine,  $\gamma$ -amine butyric acid, sarcosine, betaine, homoserine, or citrulline; or mono, di or tri- $C_{1-6}$ -alkylamino. For example, compounds wherein  $R^{10}$  is  $-\text{PO}_2(\text{OH})^-\text{M}^+$  is intended to include the monopotassium, monolithium, and monosodium salts.

5 As is appreciated by those of ordinary skill in the art are groups suitable for protecting the hydroxyl groups include trialkyl silyl, acetate, benzoate, and ether protection. See also Protective Groups in Organic Synthesis, Theodora W. Green, John Wiley and Sons 1981.

One aspect of the invention is outlined in Scheme 1 and comprises four steps.

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Scheme 1



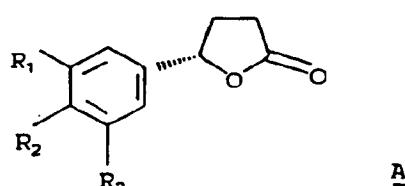
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As shown in Scheme 1, butyrolactone, Compound A, is reduced to a lactol, then silylated providing silylactol, Compound B. Compound B is then activated through treatment with silyl bromide, forming a glycosyl bromide Compound C. Coupling is subsequently achieved using an aryl copper species to stereoselectively produce the target *trans*-2,5-diaryltetrahydrofuran, Compound 2D.

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40 In one embodiment, the invention concerns a process of making compounds of the formula 2D, comprising:  
(A) contacting of a compound of the formula

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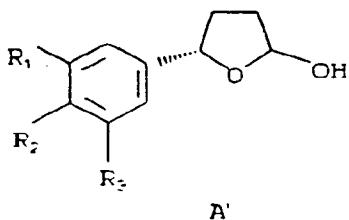


A

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in an aromatic solvent with a reducing agent to yield compound A';

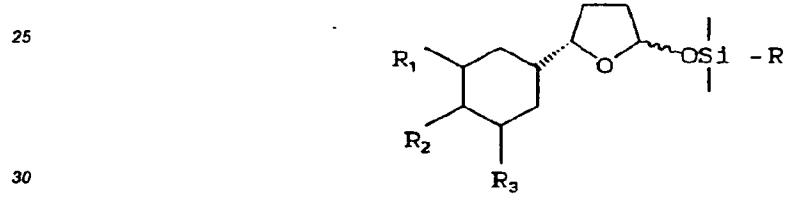
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10 For purposes of the specification, aromatic solvents include, but are not limited to, benzene, toluene and xylene, preferably toluene. Reducing agents include, but are not limited to metal hydrides such as sodium bis-methoxy, ethoxy aluminum hydride and diisobutylaluminum hydride preferably, diisobutylaluminum hydride. For complete reaction the molar ratio of reducing agents to lactone should be approximately 1 to 1, or larger; 15 preferably 1.25 to 1.. The reaction may be conducted from -80°C to -50°C, preferably -75°C to -60°C. The reaction is allowed to proceed until substantially complete in about 1 to 2 hours, typically 1.25 or 1.5 hours. The reaction can then be quenched by addition of C<sub>1-6</sub>alkanol such as methanol.

While the reaction can be carried out at up to 100 atmospheres of pressure, the reaction is preferably carried at under ambient pressure;

20 (B) contacting of compound A' with a tri C<sub>1-6</sub>alkyl chlorosilane in a second solvent and a base to yield the silyllactol Compound B;



wherein R is C<sub>1-6</sub> alkyl;

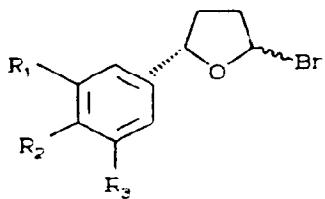
35 For purposes of this specification tri-C<sub>1-6</sub>alkylchlorosilanes include but are not limited to tri-C<sub>1-6</sub> alkyl chlorosilane wherein each alkyl group is independently defined as C<sub>1-6</sub> alkyl. Preferred is tert-butyl-dimethylchlorosilane. The second solvent includes, but is not limited to N,N-diC<sub>1-6</sub>alkyl carbonyl amide, such as N,N-dimethyl formamide (DMF) or toluene, tetrahydrofuran (THF), dichloromethane or other non-protic solvent; DMF is preferred. Nitrogen containing bases include but are not limited to pyrrole, pyridene, pyrrolidine 40 tri C<sub>1-3</sub>alkyl amino such as triethyl amine and imidazole. Imidazole is preferred for complete reaction. The molar ratio of base to compound A' should be approximately 2 to 1 or greater. A ratio of 2.2 to 1 is typical. The ratio of silane to compound A is approximately 1.1 to 1 up to 2.5 to 1; preferably 1 to 1. The reaction should be allowed to proceed until complete in approximately 1 to 3 hrs. The reaction temperature may be 0 to 80°C., preferably 25 - 30°C.

45 While the reaction can be carried out at up to 100 atmospheres of pressure, the reaction is preferably carried at under ambient pressure. The presence of oxygen is preferably minimized, such as by use of a nitrogen or other inert atmosphere.

(C) contacting of Compound B with a silyl bromide in a third solvent to yield a glycosyl bromide Compound C;

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wherein the hydroxyl groups of substituent R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are protected.

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For purposes of the specification, the third solvent includes but is not limited to etheral solvents such as diethyl ether di-n-butyl and diisopentyl ethers, anisole, cyclic ethers such as tetrahydropyran, 4-methyl-1,3-dioxane, tetrahydrofurfuryl methyl ether, ethyl ether, furan and tetrahydrofuran, or halocarbon solvents such as mono or di halo C<sub>1-4</sub>alkyl including methylene chloride. Methylene chloride is preferred. The silyl bromide includes, but is not limited to tri C<sub>1-6</sub> alkylsilyl bromide with trimethylsilyl bromide preferred for complete reaction. The molar ratio of silyl bromide to compound B should be 1 to or greater, preferably 1.1 - 1.3 to 1. The reaction is allowed to proceed until essentially complete in about 0.5 to 3 hours, typically 1.5 hrs.

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The reaction temperature is approximately -70 to -10°C, preferably -60°C.

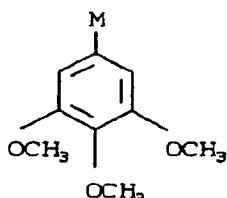
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While the reaction can be carried out at up to 100 atmospheres of pressure, the reaction is preferably carried at under ambient pressure. The presence of oxygen is preferably minimized, such as by use of a nitrogen or other inert atmosphere.

(D) contacting of Compound C with an organo metallic reagent species of the formula

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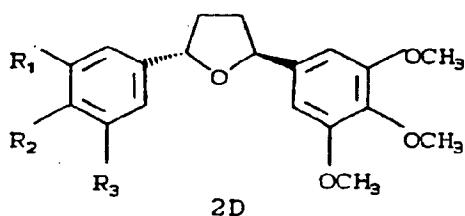
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wherein M is magnesium, aluminum, zinc, copper or lithium in a fourth solvent to yield a compound of formula 2D;

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For purposes of this specification the fourth solvent includes, but is not limited to ethers as broadly defined above; preferably THF. The organo metallic reagent includes, but is not limited to those derived from aryl Grignard reagents such as 3,4,5-trimethoxy phenylmagnesium bromide in the presence of a copper salt such as copper cyanide or dilithium tetrachlorocuprate.

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The ratio of organometallic reagent to compound C is approximately 1-1.5 to 1, preferably 1.4 to 1. The reaction is allowed to proceed until essentially complete in about .5 to 3. hours. Typically 1.0 hours. The reaction temperature is approximately -70 to -10°C, preferably -60°C

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While the reaction can be carried out at up to 100 atmospheres of pressure, the reaction is preferably carried at under ambient pressure. The presence of oxygen is preferably minimized, such as by use of a nitrogen or other inert atmosphere.

In one aspect invention described herein concerns a process for the preparation of 5-aryl-γ-butyrolactones. These compounds are critical intermediates for the synthesis of optically pure trans-2,5-diaryl tetrahydrofurans

which are potent antagonists of Platelet Activating Factor.

This invention further concerns a process of making compounds of formula 2D wherein:

R<sub>1</sub> is I, R<sub>2</sub> is OCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> and R<sub>3</sub> = CH<sub>3</sub>O;

R<sub>1</sub> = I; and R<sub>3</sub> is O (CH<sub>2</sub>)<sub>3</sub>-O-PO<sub>3</sub>H<sub>2</sub>, R<sub>2</sub> is OCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> and

R<sub>1</sub> = I; (CH<sub>2</sub>)<sub>3</sub>-OH, R<sub>2</sub> is OCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> and R<sub>3</sub> is CH<sub>3</sub>O.

The resulting compound of Formula 2D can be converted to optically pure (-)-(2S,5S)-2-(3(2-hydroxyethylsulfonyl)-4-(n-propoxy)-5-methoxyphenyl)-5-(3,4,5-trimethoxyphenyl)tetrahydrofuran, a potent PAF antagonist.

Additional optically pure PAF antagonists that can be produced include (-)-(2S,5S)-2-(3(2-oxopropylsulfonyl)-4-(n-propoxy)-5-(3-phosphopropoxy)phenyl-5-(3,4,5-trimethoxyphenyl)-tetrahydrofuran; (-)-(2S,5S)-2-(3(2-oxopropylsulfonyl)-4-(n-propoxy)-5-(3-hydroxypropoxy)phenyl-5-(3,4,5-trimethoxyphenyl)-tetrahydrofuran; and

(-)-(2S,5S)-2-(3(2-hydroxypropylsulfonyl)-4-(n-propoxy)-5-(3-hydroxypropoxy)phenyl-5-(3,4,5-trimethoxyphenyl)-tetrahydrofuran.

The compound of Formula A may be prepared according to the method of Corey et. al., *J. Am. Chem. Soc.*, (1987), 109, 7925. Preferably they are prepared as outlined in Scheme 2 and described in greater detail thereafter.

Scheme 2

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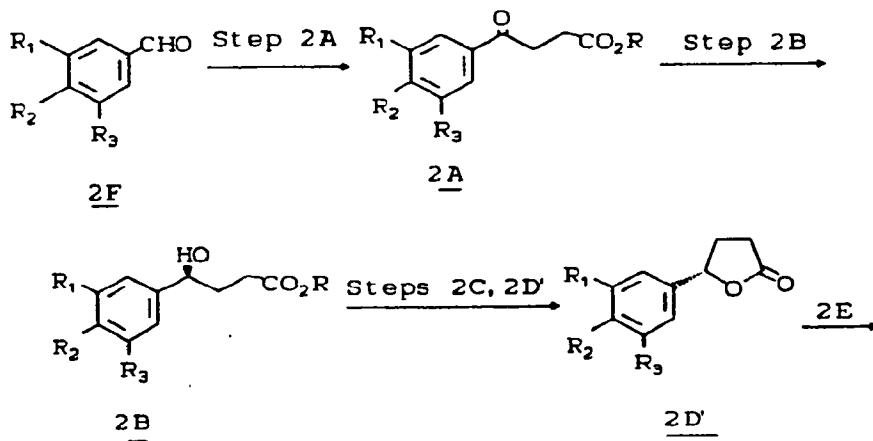
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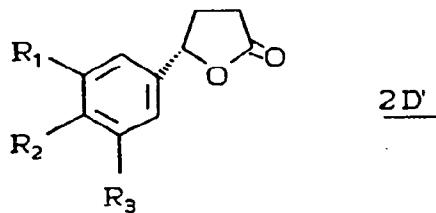


In step 2A an *in situ* Prepared acyl anion equivalent, compound 2F, which is derived from a substituted benzaldehyde is chemoselectively added to an  $\alpha,\beta$ -unsaturated ester, to yield Compound 2A. This single transformation assembles the requisite carbon framework from commercially available precursors. In step 2B an enantioselective reduction utilizes  $\beta$ -chlorodiisopinocampheyl borane in an unprecedented manner to produce an optically enriched 4-aryl-4-hydroxy-butanoate, Compound 2B. In steps 2C to 2D' conversion of Compound 2B to the title lactone Compound 2D' (also designate Compound A) is accomplished via a novel, internally assisted saponification followed by a mild acid catalyzed lactonization. Both saponification and lactonization are effected without racemization. Thereafter, controlled crystallization of Compound 2D' efficiently enriches the optical purity to greater than 99.5%.

This approach envisions a process of making Compounds of Formula 2D'

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wherein:

R<sup>1</sup> is iodide, or bromide, or

R<sup>1</sup> is S(O)<sub>n</sub>R<sub>a</sub> in which n is 0,1 or 2 and R<sub>a</sub> is selected from the group consisting of

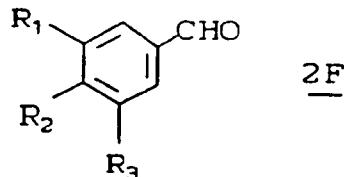
- (a) C<sub>1-6</sub>alkyl,
- (b) C<sub>2-6</sub>alkenyl,
- (c) C<sub>2-6</sub>alkynyl,
- (d) substituted C<sub>1-6</sub>alkyl wherein the substituent is selected from the group consisting of hydroxy, N-C1-4alkylamino, and N,N-C1-4di-alkylamino,
- (e) C<sub>1-6</sub>alkoxy-C<sub>1-6</sub>alkyl,
- (f) C<sub>1-6</sub>alkylcarbonyl-C<sub>1-6</sub>alkyl; and

R<sub>2</sub> is selected from the group consisting of

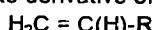
- (a) C<sub>1-12</sub>alkoxy,
- (b) C<sub>2-6</sub>alkenyloxy,
- (c) C<sub>2-6</sub>alkynyoxy,
- (d) C<sub>2-6</sub>(halo)x alkoxy wherein x is 1,2,3,4 or 5 and halo is chloro fluoro or bromo,
- (e) substituted C<sub>1-6</sub>alkoxy wherein the substituent is hydroxy,
- (f) C<sub>1-6</sub>alkoxy-C<sub>1-6</sub>alkoxy,
- (g) C<sub>1-6</sub>alkyl S(O)<sub>m</sub>-C<sub>1-6</sub>alkoxy in which m is 0, 1 or 2,
- (h) C<sub>1-6</sub>alkyloxsulfonyl-C<sub>1-6</sub>alkoxy,
- (i) C<sub>1-6</sub>alkyl carbonyl-C<sub>1-6</sub>alkoxy,
- (j) phenyl-C<sub>1-6</sub>alkoxy,
- (k) azido-C<sub>1-6</sub>alkoxy,
- (l) cyano-C<sub>1-6</sub>alkoxy,
- (m) C<sub>1-6</sub> alkylS(O)<sub>m</sub>-C<sub>1-6</sub>alkoxy,
- (n) N-substituted, or N,N-disubstituted amino-C<sub>1-6</sub> alkyloxy wherein the substituents are each individually C<sub>1-6</sub>alkyl;

R<sub>3</sub> is selected from the group consisting of

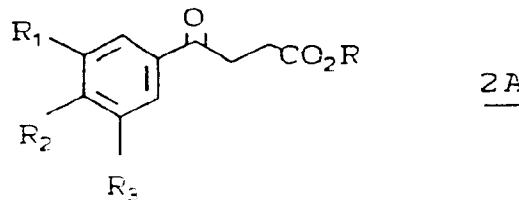
- (a) C<sub>1-6</sub>alkoxy,
- (b) substituted C<sub>1-6</sub>alkoxy wherein the substituent is selected from the group consisting of hydroxy, N-C<sub>1-4</sub>alkylamino, N,N-diC<sub>1-4</sub>alkylamino,
- (c) -O-C<sub>1-6</sub>alkyl-0-R<sup>10</sup>, wherein R<sup>10</sup> is
  - (1) -PO<sub>2</sub>(OH)<sup>-</sup> M<sup>+</sup> wherein M<sup>+</sup> is a pharmaceutically acceptable cation,
  - (2) -C(O)(CH<sub>2</sub>)<sub>2</sub>-CO<sub>2</sub><sup>-</sup>M<sup>+</sup>, or
  - (3) -SO<sub>3</sub><sup>-</sup>M<sup>+</sup>,
- (d) C<sub>1-6</sub>alkylcarbonyl-C<sub>1-6</sub>alkoxy,
- (e) C<sub>1-6</sub>alkoxyaminocarbonyloxy,
- (f) halophenylC<sub>1-6</sub>alkoxy, and
- (g) C<sub>1-6</sub>carboxyalkoxy, comprising: (2A) Contacting, in the substantial absence of oxygen, and in the presence of a catalyst compound of formula 2F



wherein R<sub>1</sub> is iodide or bromide, with an acrylate derivative of formula



55 wherein R is CO<sub>2</sub>Et, CO<sub>2</sub>Me, CO<sub>2</sub>CH<sub>2</sub>Ph, CO<sub>2</sub>CH<sub>2</sub>CHCH<sub>2</sub>, CO<sub>2</sub>Ph, CO<sub>2</sub>-t-C<sub>4</sub>H<sub>9</sub> or CN, to yield a compound of formula 2A;



10 Preferably, contacting step 2A is carried out in two stages. The first stage comprises degassing a solution of compound 2F in the first solvent, followed by addition of a catalytic amount of alkali metal cyanide to the solution of compound 2F in the first solvent. Degassing may conveniently be accomplished by bubbling nitrogen gas through the solution for 10 minutes under ambient conditions. The cyanide is then added and the reagents are stirred for about 10 to 100 minutes. 30 minutes under constant stirring has proven quite satisfactory.

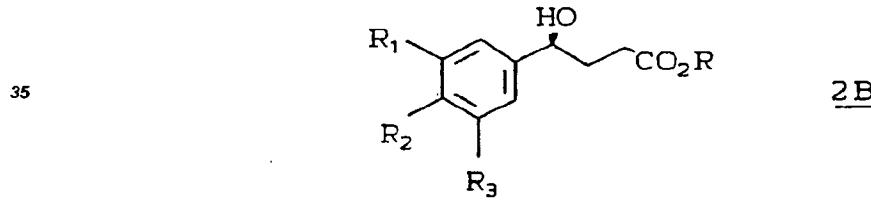
15 While the first stage may be carried out at up to 100 atmospheres, this stage is preferably carried out at ambient pressure. Temperature can range from 20 to 30°C, but is preferably at about 25°C. The ratio of alkali metal cyanide to compound 2F is 0.1 to 0.3 moles per 100 moles, most preferably 0.25 mole.

20 Contacting step 2A is then completed by direct addition of the acrylate derivative, preferably over a 50 to 60 minute period, at from 0 to 25°C.

25 For purposes of this specification, the first solvent includes, but is not limited to, such solvents as mono or di C<sub>1</sub>-<sub>6</sub> alkyl amide derivatives such as dimethylformamide (DMF); di-C<sub>1</sub>-<sub>6</sub> alkyl sulfoxide, such as methylsulfoxide or aqueous C<sub>1</sub>-<sub>6</sub> alcohol, such as ethanol, most preferably DMF. The alkali metal cyanide is a cyanide such as sodium, potassium or lithium cyanide, preferably sodium cyanide.

30 The acrylate derivative, is preferably a sterically hindered acrylate, such as CO<sub>2</sub>-t-C<sub>4</sub>C<sub>9</sub>. The selected acrylate is preferably added gradually over 1 hour to provide the desired  $\gamma$ -keto ester of formula 2A in a yield of approximately 80% (for R = CO<sub>2</sub>-t-C<sub>4</sub>H<sub>9</sub>, 80%). Critical to reaction success was the discovery that oxygen exclusion is a requirement. In its presence, oxidative decomposition leading to by-products depress the yield significantly;

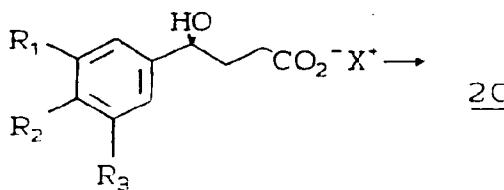
35 (2B) Contacting the compound of formula 2A in an etheral solvent with  $\beta$ -chlorodiisopinocampheyl borane to yield a compound of formula 2B



45 For purposes of this specification, etheral solvents include, but are not limited to ethers such as diethyl ether di-n-butyl and diisopentyl ethers, anisole, cyclic esters such as tetrahydropyran, 4-methyl-1,3-dioxane, dihydropyran, tetrahydrofurfuryl methyl ether, ethyl ether, furan and 2-ethoxytetrahydrofuran, most preferably tetrahydrofuran.

50 The reaction can be conducted at -25 to 25°C preferably at 0 to 5°C. The reaction is allowed to proceed until essentially complete in about 1 to 100 hours, preferably 18 hours. While the pretreatment may be carried out at up to 100 atmospheres, the reaction is preferably carried out at ambient pressure. The  $\gamma$ -hydroxy butanoate derivative compound 2B is provided in typically 80-90% yields with an enantiomeric excess (ee) of 92%. Use of the (-)-chloroborane enantiomer provides the 4S-alcohol while the (+)-chloroborane enantiomer yields the 4R-alcohol. Thus both enantiomers of 2B are accessible by this invention. The reaction of  $\gamma$ -keto esters with this reducing agent is highly unprecedented as prior art indicates that  $\beta$ -keto-esters are not suitable substrates.

55 (2C) Contacting compound 2B in a medium containing alcohol in an etheral solvent with an alkali metal hydroxide to yield a compound of formula 2C.



wherein X is an alkali metal selected from the group consisting of Sodium, Potassium and Lithium.

10 For purposes of this specification, alcohol includes, but is not limited to C<sub>1-6</sub> alkanol, preferably ethanol. As before, sodium hydroxide is the preferred alkali metal hydroxide. For purposes of this specification, ethereal solvents include, but are not limited to ethers such as diethyl ether di-*n*-butyl and diisopentyl ethers, anisole, cyclic esters such as tetrahydropyran, 4-methyl-1,3-dioxane, dihydropyran, tetrahydrofurfuryl methyl ether, ethyl ether, furan and 2-ethoxytetrahydrofuran, most preferably tetrahydrofuran. For complete saponification, the molar ratio of alkali metal hydroxide to compound 2C should be at least 1 to 1, preferably 1.5 to 1 or greater. The time, temperature and pressure of the reaction are not considered critical. The reaction can be conducted at -25 to 50°C, preferably at 25°C. The reaction is allowed to proceed until essentially complete in about 20 to 200 minutes, preferably 75 minutes. While the reaction may be carried out at up to 100 atmospheres, the reaction is preferably carried out at ambient pressure.

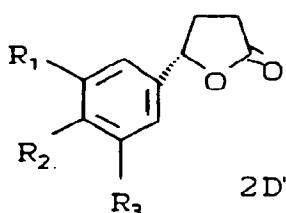
20 Highlighting this step is the intramolecular assistance provided by the  $\gamma$ -hydroxyl moiety in compound 2B which facilitates removal of the R-oxy group under basic conditions. Normally recommended acid catalyzed procedures for hydrolysis of the R ester would likely result in significant racemization of this substrate. Saponification yields compound 2C as a free acid salt, readily extractable into water and consequently easily separated from the neutral pinanyl by-products resulting from the chiral reduction step.

25 Thereafter the acid salt of compound 2C can be converted to the acid by any of the conventional means known in the art. Our preferred procedure is described in the Example and incorporates partial purification of the resulting acid.

(2D) Contacting the free acid of compound 2C in a second solvent with pyridinium para-toluene sulfonate to yield a compound of formula 2D'.

30

35



40 For purposes of this specification the second solvent includes, but is not limited to, an ethereal solvent, as defined above, or a C<sub>6-10</sub> linear, branched or cyclic hydrocarbon solvent. Toluene is preferred. The time, temperature and pressure of the reaction are not considered critical. The reaction can be conducted at 50 to 80°C, preferably at 70°C. The reaction is allowed to proceed until essentially complete in about 20 to 200 minutes, preferably 90 minutes. While the reaction may be carried out at up to 100 atmospheres, the reaction is preferably carried out at under ambient pressure in a nitrogen atmosphere.

45 Significantly, racemization does not occur, even with highly electron rich substrates.

50 The 80 to 95% optically pure product can be optically enriched to greater than 99.5% enantiomeric excess by controlled crystallization from ethyl acetate, isopropyl acetate, ethanol, methanol, or solvent mixtures of a hydrocarbon solvent such as hexanes, cyclohexane and esters such as ethyl acetate, isopropyl acetate or ethers such as methyl t-butyl ether. Preferably the optically enriched product is crystallized from an ethyl acetate/hexane mixture in a 1:6 ratio v/v at -10°C to 20°C. This provides 99.5% ee pure compound 2D.

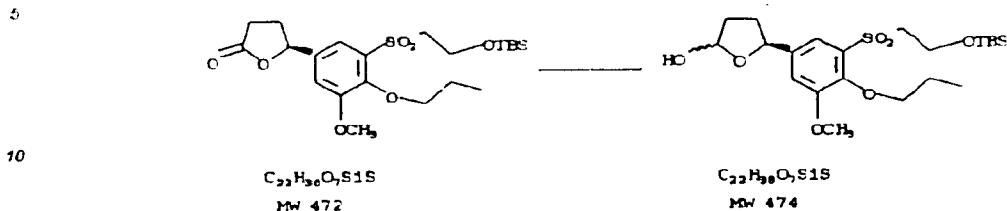
The following examples illustrate the present invention and as such are not to be considered as limiting the invention set forth in the claims appended thereto.

The starting materials are either known or described hereunder.

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### EXEMPE 1

**Step A:** 4-[3-Methoxy-4-n-propyloxy-5-(2'-t-butyl-dimethylsiloxyethylsulfonyl)phenyl]-4-butyrolactol

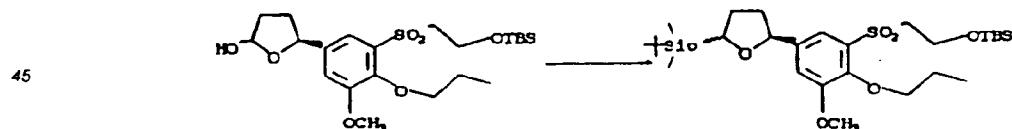


15	<u>Materials</u>	<u>Amt</u>	<u>Mols</u>	<u>MW</u>
20	4[3-Methoxy-4-n-propyloxy-5-(2'-t-butyl-dimethylsiloxyethylsulfonyl)phenyl]-4-butylactone	1.607 Kg	3.405	472
	Diisobutylaluminum hydride 1.5 M in toluene	3.5 L	5.25	1.5M
	Methanol (d=0.791)	1.5 L	37.08	32
25	Potassium sodium tartrate tetrahydrate	12 L		281.2
	Ethyl Acetate	12 L		
	Toluene	13 L		

30 To a solution of the lactone (1.607 kg, 3.405 mole) in sieve dried toluene (13 L) at -72°C is added a 1.5 M toluene solution of diisobutylaluminum hydride (3.50 L, 5.25 mole) dropwise over 1.25 hours maintaining an internal temperature of <-65°C. The mixture is stirred at -70°C for 1.0 hours.

The reaction is quenched through the slow addition of methanol (1.5 L) at -70°C then the mixture is warmed to -20°C. Saturated Rochelle's salt (12 L) is added over 0.5 hours keeping the temperature <10°C and the mixture then stirred at 5°C for 1.5 hours, then the two phases separated. The aqueous layer is extracted with ethyl acetate (12 L). The organic phase is washed with DI water (2 X 8.0 L) and with saturated aqueous sodium chloride (10 L). The organic extracts are concentrated in vacuo to provide 1.799 Kg of the lactol as a light yellow oil. HPLC assay indicated this product to be 87 wt % pure (97% yield). The lactol is suitable for use without further purification.

40 Step B: 5[3-Methoxy-4-n-propyloxy-5-(2'-t-butylidemethylsiloxyethylsulfonyl)phenyl]-1-(t-butylidemethylsiloxy)-butyrolactol

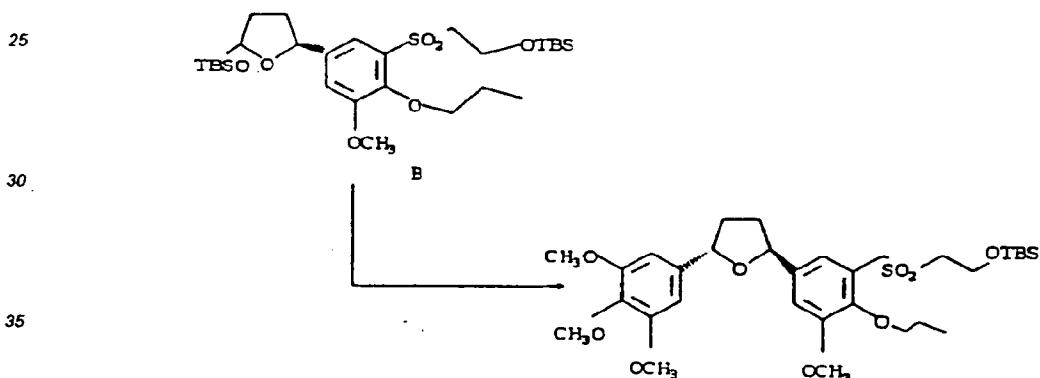


	<u>Materials</u>	<u>Amt</u>	<u>Mole</u>	<u>MW</u>
5	5[3-Methoxy-4-n-propyloxy-5-(2'-t-butyl- dimethylsiloxyethylsulfonyl)phenyl]-1- (t-butyldimethylsiloxy)-1-butyrolactol	1.522 Kg	3.211	474
10	imidazole	0.48 Kg	7.059	68
15	t-Butyldimethylsilylchloride	0.53 Kg	3.533	150
20	Dimethylformamide	3.0 L		

To a solution of the lactol (1.522 Kg, 3.211 mole) in sieve dried DMF at 25°C, under N<sub>2</sub>, was added imidazole (0.48 Kg, 7.059 mole), followed by t-butyldimethylsilylchloride (0.53 Kg, 3.533 mole). The internal temperature rises to +34°C within 1/2 hour, then cools to 25°C. Stir at 25°C, under N<sub>2</sub> for 3 hours. The reaction was diluted with EtOAc (20 liter), washed with H<sub>2</sub>O (3 X 10 L) followed by brine (10 L). The organics were concentrated to afford 2.170 Kg of a yellow oil. 300 MHz NMR is consistent for silyl hemiacetal.

HPLC assay indicated this product to be 87.5% pure (100% yield). This material is suitable for use without further purification.

Step C: Preparation of 1-tert-butyldimethylsiloxy-2-((2-methoxy-2-propyloxy-5-(tetrahydro-5-(3,4,5-trime-thoxyphenyl)-2-furanyl)phenyl-sulfonyl-trans-(-)-ethane



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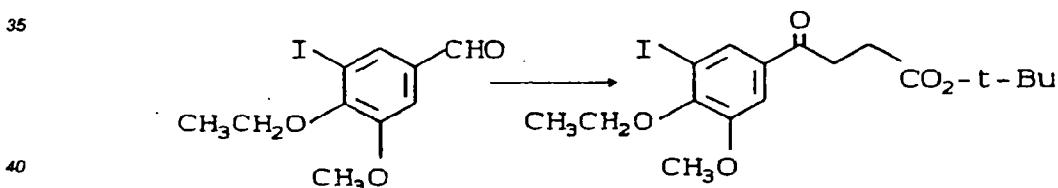
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Materials

5	5-[3-Methoxy-4-n-propyloxy-5-			
10	(2'-t-butyldimethylsiloxyethyl)-			
15	sulfonyl)phenyl]-1-(t-butyldi-			
20	methylsiloxy)-butyrolactol (B)	0.829 Kg	1.409 mole	
25	TMS-Br	0.232 L	1.759 mole	
30	Li <sub>2</sub> CuCl <sub>4</sub> /THF 0.5M	0.060 L	0.030 mole	
35	3,4,5-trimethoxyphenyl			
40	magnesium bromide (0.9M in THF)	2.25 L	2.025 mole	
45	CH <sub>2</sub> Cl <sub>2</sub>	6.0 L		
50	Ethyl Acetate	13 L		

The silyl ether B (0.829 Kg, 1.409 mole) was dissolved in CH<sub>2</sub>Cl<sub>2</sub>, under N<sub>2</sub>. The mixture was cooled to -60°C and then neat trimethylsilyl bromide (0.232 L, 1.759 mole) was added. The mixture was stirred at -60°C for 1.5 hours. In a separate flask containing 3,4,5-trimethoxyphenylmagnesium bromide (0.9 M, 2.5L, 2.025 mole), at 0°C under N<sub>2</sub> was added a THF solution of Li<sub>2</sub>CuCl<sub>4</sub> (0.060 mL, 0.030 mole).

To the glycosyl bromide at -60°C was transferred the solution of organometallic. After complete addition, the reaction was stirred at -60°C for 1.0 hours. It was quenched at -60°C by addition of 10 L of saturated NH<sub>4</sub>Cl/NH<sub>4</sub>OH (10:1 v/v), and H<sub>2</sub>O (5 L). Allow to stir without external cooling for 0.5 hours. After separating the organic layer, the aqueous layer was extracted with EtOAc (10 L) and the combined organics were washed with brine (8 L). The resulting clear, homogeneous organic layer was concentrated to afford 1.178 Kg of a red oil. Analysis of the crude reaction mixture by HPLC assay showed 0.754 Kg (86%) of the title compound.

EXAMPLE 2Preparation of tert.-butyl 3[3-methoxy-4-n-[propyloxy-5-iodobenzoyl]propionate

	<u>Materials</u>	<u>Amt</u>	<u>Moles</u>	<u>MW</u>
45	3-Methoxy-4-n-propyloxy-5-iodobenzaldehyde (0.47M soln. in DMF)	100 mL	0.0469	320
50	Dimethyl formamide	32 mL		
55	Sodium cyanide	0.574 g	0.0117	49
60	t-Butyl acrylate	5.40 g	0.0422	128
65	Ethyl acetate	150 mL		
70	Saturated aq. NaCl	65 mL		
75	Saturated aq. NaHCO <sub>3</sub>	75 mL		
80	Water	195 mL		

## EP 0 465 122 A1

A 3-methoxy-4-n-propyloxy-5-iodobenzaldehyde solution in DMF (100 mL, 15 g, 0.0469 mole) (KF = 640  $\mu$ g/mL) is degassed by bubbling  $N_2$  through the solution for 10 minutes. Sodium cyanide (0.574 g, 0.0117 mole) is added in one portion and the mixture is stirred for 30 minutes under  $N_2$ . The temperature rises from 24°C to 28°C.

5 After a total age time of 45 minutes, the solution is cooled to 18°C. A solution of t-butyl acrylate (6.17 mL, 5.40 g, 0.0422 mol) in sieve-dried deoxygenated dimethylformamide (32 mL) (KF = 170  $\mu$ g/mL) is added over 50 minutes at 18°C.

10 The reaction is aged for 30 minutes after complete acrylate addition. Ethyl acetate (75 mL), distilled water (45 mL) and saturated aqueous sodium chloride (45 mL) are added to the reaction mixture and the layers are separated.

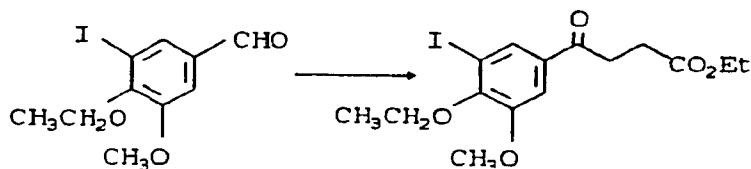
15 The aqueous phase is back extracted with ethyl acetate (75 mL). The combined organic extracts are washed with saturated aqueous sodium bicarbonate (75 mL), followed by 2 x 75 mL of distilled  $H_2O$  and 10 mL of saturated aqueous sodium chloride. The ethyl acetate layer is dried over sodium sulfate, filtered and concentrated in vacuo to give the desired ketoester in an 85% yield by HPLC assay.

## EXAMPLE 3

Ethyl 3[3-methoxy-4-n-propyloxy-5-iodobenzoyl] propionate

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	<u>Materials</u>	<u>Amt</u>	<u>Moles</u>	<u>MW</u>
	3-Methoxy-4-n-propyloxy-5-iodobenzaldehyde	3.67 kg	11.46	320.05
	Dimethyl formamide	28.4 L		
	Sodium cyanide	140.7 g	2.88	49
35	Ethyl acrylate	1.15 k	11.45	100
	Ethyl acetate	36.0 L		

40 3-Methoxy-4-n-propyloxy-5-iodobenzaldehyde (4.021 kg, 3.67 kg/HPLC assay 11.46 mole) is dissolved in dry dimethylformamide (KF - 420  $\mu$ g/ml)(21.0 L) under  $N_2$ . Sodium cyanide (140.7 g, 2.88 mole) is added in one portion and the mixture stirred for 45 minutes under nitrogen. The temperature rises from 18°C to 22°C during the age.

45 After a total age time of 1.5 hours, the batch is cooled to 20°C. A solution of ethyl acrylate (1.234 L, 1.15 kg, 11.45 moles) in dry dimethylformamide (7.4 L) is added over 2 1/3 hours at 20-23°C [cooling as needed].

50 Ethyl acetate (17.9 L), cold distilled water (11.4 L) and saturated aqueous sodium chloride solution (11.4 L) are added to the reaction mixture with stirring and the phases separated. The organic phase is washed with saturated aqueous sodium chloride solution (11.4 L). The combined aqueous layers are back-extracted with ethyl acetate (17.6 L). The combined organic extracts are washed with 2 x 17.9 L of distilled water. An emulsion which may form in the final wash can be broken through addition of 2.7 L of saturated sodium chloride solution. The ethyl acetate layer is dried over anhydrous sodium sulfate, filtered and concentrated in vacuo to give 4.462 kg of the title compound as an oil.

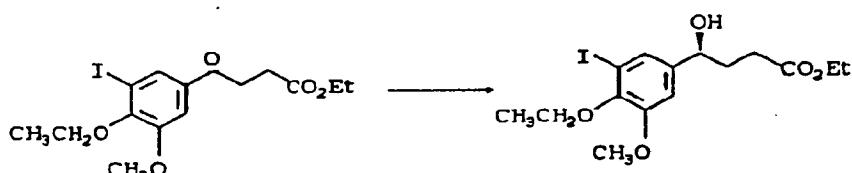
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EXAMPLE 4tert-Butyl 4[3-methoxy-4-n-propyloxy-5-iodophenyl]-4S-hydroxybutyrate

	<u>Materials</u>	<u>Amt</u>	<u>Moles</u>	<u>MW</u>
15	t-Butyl 3[3-methoxy-4-n-propyl-5-iodobenzoyl]propionate	9.3 g	0.0219	448
	(-)- $\beta$ -Chlorodiisopinocampheylborane	11.9 g	0.0372	320
20	Tetrahydrofuran	20 mL		
	30% Hydrogen Peroxide	10 mL		
	Water	100 mL		
25	Ethyl acetate	75 mL		
	Saturated aqueous sodium bicarbonate	60 mL		
	Saturated aqueous sodium chloride	30 mL		

30 (-)- $\beta$ -Chlorodiisopinocampheylborane (11.9 g, 0.0372 mol) is dissolved in sieve-dried tetrahydrofuran (10 mL, KF = 74  $\mu$ g/mL) under nitrogen. The solution is cooled to 0-5°C in an ice bath. Meanwhile, a solution of crude t-butyl 3[3-methoxy-4-n-propyl-5-benzoyl]propionate (10.0 g, 9.3 g by HPLC assay, 0.0219 mol) in dry tetrahydrofuran (10 mL) is prepared. The solution is added dropwise at 0-5°C to the chiral reducing agent under nitrogen. The reaction is stirred at 0°C and monitored by HPLC.

35 The reaction is quenched by dropwise addition of distilled water (7 mL) at 5-10°C. 30% Hydrogen peroxide (10 mL) is slowly added maintaining the temperature at 10-20°C. The mixture is added to distilled water (30 mL) and ethyl acetate (75 mL). The layers are separated and the organic phase is washed with 30 mL of distilled water, 2 x 30 mL of half-saturated aqueous sodium chloride. The organic phase is dried over anhydrous sodium sulfate, filtered and concentrated in vacuo to provide the title compound as a yellow oil (18.2 g).

EXAMPLE 5Ethyl 4[3-methoxy-4-n-propyloxy-5-iodophenyl]-4S-hydroxybutyrate

	<u>Materials</u>	<u>Amt</u>	<u>Moles</u>	<u>MW</u>
5	Ethyl 3[3-methoxy-4-n-propyl-5-benzoyl]-propionate	1.39 kg	3.31	420.05
	(-) $\beta$ -Chlorodiisopinocamphenylborane	1.9 kg	5.92	320.76
	Tetrahydrofuran	2.86 L		
10	30% Hydrogen peroxide	960 ml		
	Ethyl acetate	14.0 L.		
	Sodium bicarbonate			

15 (-) $\beta$ -Chlorodiisopinocamphenylborane (1.7 kg, 5.3 moles) is dissolved in sieve-dried tetrahydrofuran (1.43 L,  $KF = 335 \mu\text{g/ml}$ ) under nitrogen at 25°C. The solution is cooled to 0-5°C in an ice/methanol bath. Meanwhile, a solution of crude ethyl 3[3-methoxy-4-n-propyloxy-5-iodobenzoyl] propionate (1.82 kg, 1.39 kg HPLC assay basis, 3.31 moles) in dry tetrahydrofuran (1.43 L) is prepared. Anhydrous sodium sulfate is used to assure dryness of the solution. The solution is filtered and added over 1.0 hour at 0-5°C to the chiral reducing agent under nitrogen. The batch is stirred overnight at 0-5°C.

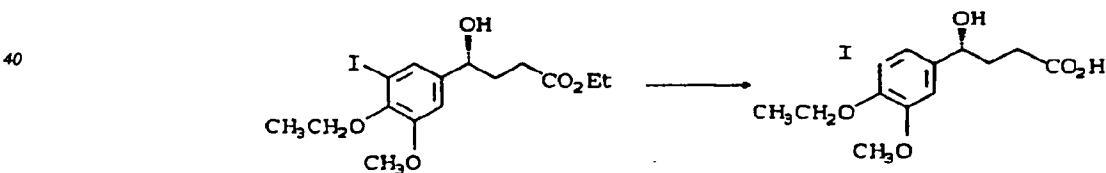
20 Additional chiral reducing agent (200 g, 0.629 mole) is charged at 0-5°C and the mixture aged for another 24 hours.

HPLC now shows 79.4% product and 0.5% s.m.; the estimated extent of reaction is now 99.7%.

25 The reaction is quenched with distilled water (477 ml) at 5-10°C over 40 minutes [dry ice/methanol cooling]. 30% Hydrogen peroxide (960 ml) is added over 1.25 hour at 15-20°C. The mixture is added to a 10-gallon extractor containing distilled water (4.0 L) and ethyl acetate (11.4 L) with stirring. The reaction vessel is rinsed with distilled water (1.77 L) and ethyl acetate (2.0 L). The layers are separated and the organic phase is washed with 5.77 L of distilled water, 2 x 5.77 liters of 10% sodium bicarbonate solution and 2 x 5.77 liters of half-saturated sodium chloride solution. The batch was dried over anhydrous sodium sulfate, filtered and concentrated in vacuo to an oil. The residue is diluted with tetrahydrofuran (2 L) and again concentrated to give 3.32 kg of hydroxy-ester as an oil. This product is hydrolyzed without further purification.

#### EXAMPLE 6

##### 4[3-Methoxy-4-n-propyloxy-5-iodophenyl]-4-hydroxybutyric acid



	<u>Materials</u>	<u>Amt</u>	<u>Moles</u>	<u>MW</u>
50	Ethyl 4[3-methoxy-4-n-propyloxy-5-iodophenyl]-4-hydroxybutyrate	3.32 kg	3.31	422.05
			(theor)	
	Sodium hydroxide	230 g	5.75	40
	Tetrahydrofuran	5.75 L		
55	Methyl-t-butyl ether	58.0 L		
	Hydrochloric acid	875 ml	1.01	11.6M.

The crude hydroxyester (3.32 kg, 3.31 moles-theor.) is dissolved in tetrahydrofuran (5.75 L) and the solution is added to 1M sodium hydroxide solution (5.75 L). The batch is aged at 25°C for 1.25 hours.

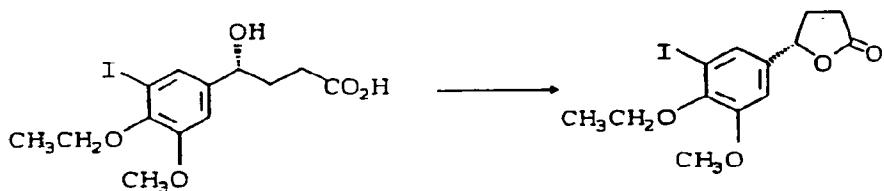
10 10% Aqueous sodium bicarbonate solution (3.83L) then methyl t-butyl ether [MTBE] (11.5 L) are added. The layers are separated [aqueous pH=9.2] and the aqueous layer extracted with another 2 x 11.5 L of MTBE. 5 The alkaline aqueous layer is acidified to pH=2.0 with 6N aqueous hydrochloric acid (1.75 L) and the  $\gamma$ -hydroxy acid product extracted into 2 x 11.5 L of MTBE. The organic phase is dried over anhydrous sodium sulfate, filtered and concentrated in vacuo to an oil. The residue is flushed with toluene (2.0 L) and again concentrated to give 1.35 kg of crude  $\gamma$ -hydroxy acid as an oil. This is used without purification in the lactonization step.

10 EXAMPLE 7

4[3-Methoxy-4-n-propyloxy-5-iodophenyl]-4S-butyrolactone

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<u>Materials</u>	<u>Amt</u>	<u>Mole</u>	<u>MW</u>
4[3-Methoxy-4-n-propyloxy-5-iodophenyl]-4S-hydroxybutyric acid	1.35 kg	3.31 (theor)	392
Pyridinium-p-toluenesulfonate	12.1 g	0.048	251.3
Toluene	11. L		

The crude  $\gamma$ -hydroxyacid (1.35 Kg) is dissolved in toluene (11.5 L) and charged to a 50 L R.B. flask. Pyridinium-p-toluenesulfonate [PPTS] (12.1 g, 0.048 mole) is added and the mixture heated to 70°C under N<sub>2</sub>. 35 The reaction mixture is placed under vacuum [ca. 20 inches] and water is azeotropically removed. The batch is aged at 70°C for 1.5 hours.

The mixture is concentrated in vacuo, to a total volume of 4.0 L and transferred to a large Buchi Rotovapor, concentrated to an oil, flushed with 2 x 2.0 liters of ethyl acetate and again concentrated to an oil which solidifies as a waxy solid. This residue is dissolved in 900 ml of ethyl acetate, filtered to remove a small amount of insoluble material (1.2 g) and the solids washed with 250 ml of hexanes. Hexanes (2.0 L) are added while stirring, and seeded with 100 mg of lactone. Additional hexanes (3.07 L) are added and the batch is stirred overnight at 0-5°C to complete the crystallization. The solids are filtered, washed with 3 x 260 ml of cold hexanes/ethyl acetate [4/1] and dried in vacuo at 25°C, to yield, 947.6 g of an off-white solid. This material is of acceptable quality for continued processing.

45 NMR [4 mg lactone + 40 mg (S)-(+)-2,2,2,-trifluoro-1-(9-anthryl) ethanol in CD<sub>2</sub>Cl<sub>2</sub>] showed >99.5% ee.

EXAMPLE 8

Lactone Recrystallization

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In cases in which the initially isolated lactone is not suitable for continued processing [<99.5% ee]. The following recrystallization procedure is employed.

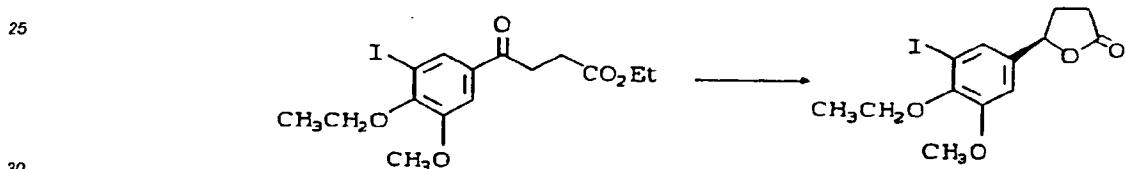
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Materials

5.	4[3-Methoxy-4-n-propyloxy-5-iodophenyl]-4S-butyrolactone	1.656 kg
	Ethyl acetate	1.9 L
	Hexanes	7.2 L

10 Crude lactone [1.656 kg] is dissolved in ethyl acetate (1.65 L) at 45°C and filtered from insolubles [ca. 5 g]. The insolubles are washed with 250 ml of ethyl acetate. Hexanes (1.8 L) are added to the combined filtrate plus washes and seeded with 100 mg of lactone. Additional hexanes (5.46 L) are added and the batch allowed to crystallize for one half hour at 25°C. Crystallization is completed by aging overnight in the cold room. The solids are filtered, washed with 3 x 500 ml of cold hexanes/ethyl acetate [4/1] and dried in vacuo at 25°C to yield 1.377 kg [83.2%].

15 NMR [4 mg lactone + 40 mg (S)-(+)2,2,2,-trifluoro-1(9-anthryl)ethanol in CD<sub>2</sub>Cl<sub>2</sub>] showed >99.5% ee; HPLC (wt. %) was 98.9%

EXAMPLE 920 4[3-Methoxy-4-n-propyloxy-5-iodophenyl]-4R-butyrolactoneMaterials

	Amt	Moles	MW
35			
Ethyl 3[3-methoxy-4-n-propyl-5-iodo			
benzoyl]-propionate	30.08 g	0.073	420.05
(+)-β-chlorodiisopinocampheyborane	38.56 g	0.121	320.76
40			
Tetrahydrofuran	125 ml		

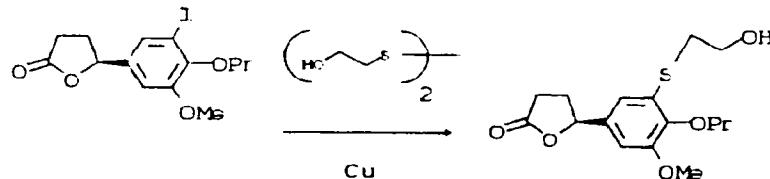
45 A solution of (30.80 g, 73 mmol) of keto-ester in 25 ml of THF was added dropwise to a solution of (+)-β-chlorodiisopinocampheyborane (32.5 g, 0.102 mole) in 100 ml of THF at 0°C. After 15 hours at 0-5° an additional 6.0 g (0.019 g) of reducing agent was added and the reaction continued for an additional 27 hours. The reaction was quenched by dropwise addition of 20 ml of water followed by slow addition of 50 ml of 30% hydrogen peroxide over 30 minutes keeping the temperature <15°C. An additional 50 ml of water was added and the mixture was extracted with 400 ml of ethyl acetate. The organics were washed with water and saturated sodium bicarbonate, dried over sodium sulfate and concentrated in vacuo. The crude mixture was stirred at room temperature in 500 ml of THF and 110 ml of 1M sodium hydroxide for 1 hour. The mixture was diluted with 200 ml of saturated sodium bicarbonate and washed with 200 ml of diethyl ether. The aqueous phase was acidified with 6M HCl to pH 3 and extracted with diethyl ether. The organic layer was dried over sodium sulfate and concentrated in vacuo to yield 18.37 g of hydroxy acid. The crude hydroxy-acid was dissolved in 250 ml of toluene and 0.179 of pyridinium p-toluenesulfonic acid was added. The mixture was heated to 60°C for 1 hour and concentrated in vacuo to a light yellow solid. Crystallization of the solid from ethyl acetate:hexane yielded 11.5 g of the 4R-butyrolactone which was >99% optically pure as determined by NMR using (S)(+)-2,2,2,-trifluoro-1(9-anthryl)ethanol as a chiral solvating agent.

## EXAMPLE 10

## 4-[3-Methoxy-4-n-propyl-5-(2'-hydroxyethylthio)phenyl]-4-butyrolactone

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	<u>Materials</u>	<u>Amount</u>	<u>Moles</u>	<u>MW</u>
20	4-[3-Methoxy-4-n-propyl-5-iodophenyl]-4-butyrolactone	2.0 g	5.33 mmol	376.0
25	Copper Powder (99%)	0.51 g	7.99 mmol	63.5
25	2-Hydroxyethyl disulfide(95%)	0.66 g	4.264 mmol	154.2
30	Dimethylformamide	15 ml		
30	Ethyl acetate	65 ml		

35 Iodolactone (2.0 g, 5.33 mmol) is dissolved in dimethylformamide (15 ml KF <200 µg/ml) at ambient temperature. Copper powder (0.51 g, 7.995 mmol) and then 2-hydroxyethyl disulfide (0.66 g, 4.264 mmol) is added to the solution. The mixture is heated to 108°C for 22 hours. HPLC analysis [C-8, acetonitrile: water:phosphoric acid 60:40:0.1, 254 nm] shows no starting iodide and 3-5% formate ester byproduct.

40

Iodolactone: retention times = 8.8 min.

Formate ester: " " = 5.0 min.

Sulfide: " " = 3.2 min.

45

45 The mixture is cooled to ambient temperature and 40 mL of ethyl acetate is added. The solution is stirred for 15 minutes and filtered through a celite pad. The addition of ethyl acetate prior to filtration greatly improves phase separation. The cake is washed with 25 ml of ethyl acetate. The combined organic extracts are washed with 3 x 40 ml of an ammonium chloride:ammonium hydroxide solution, followed by 40 ml of water. The ammonium chloride:ammonium hydroxide solution is prepared by adding approximately 65 ml of ammonium hydroxide solution (30%) to 300 ml of saturated aqueous ammonium chloride to a pH of 9.0. A pH range of 8.5-10.0 for this work has been determined to be satisfactory although pH 9.0 is favorable.

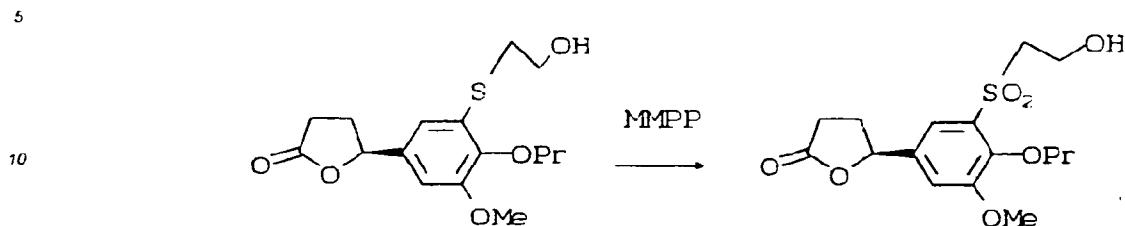
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The organic extract is concentrated in vacuo to a volume of 4 ml. The solution is diluted with 2 x 20 ml of acetonitrile and reconcentrated to ~4 mL. The acetonitrile solution is used directly for the next step.

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HPLC assay typically shows an 85-90% yield. <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) δ 6.89 (d, J=1.8Hz, 1H), 6.76 (d, J=1.8Hz, 1H), 5.40 (dd, J=6.0, 8.2Hz, 1H), 3.95 (t, J=6.8Hz, 2H), 3.83 (s, 3H), 3.66 (q, J=6.0Hz, 2H), 3.04 (t, J=5.9Hz, 2H), 2.69-2.59 (m, 4H), 2.20-2.13 (m, 1H), 1.81 (sextet, J=7.1Hz, 2H), 1.03 (t, J=7.4Hz, 3H).

<sup>13</sup>C-NMR (300 MHz, CDCl<sub>3</sub>) δ 176.8, 153.3, 147.5, 135.5, 129.8, 119.6, 108.4, 80.9, 75.2, 60.3, 56.1, 36.5, 31.0, 29.1, 23.5, 10.5.

EXAMPLE 114-[3-Methoxy-4-n-propyl-5-(2'-hydroxyethylsulfonyl) phenyl]-4-butyrolactone

	<u>Materials</u>	<u>Amount</u>	<u>Moles</u>	<u>MW</u>
20	4-[3-Methoxy-4-n-propyl-5-(2'-hydroxyethylsulfonyl)phenyl]-4-butyrolactone	5.00 g	15.3 mmol	326.0
25	Monoperoxyphthalic acid magnesium salt	13.66 g	27.6 mmol	494.6
30	Acetonitrile	27 ml		
35	Water	40 ml		
40	Saturated NaHCO <sub>3</sub>	195 ml		
45	5% NaCl	50 ml		
	Ethyl Acetate	110 ml		
	DMF	100 ml		

Monoperoxyphthalic acid magnesium salt (13.66 g, 27.6 mmol) was suspended in 40 ml of water at ambient temperature. A solution of sulfide (5.0 g, 15.3 mmol) in 27 ml of acetonitrile was added dropwise over 15 minutes keeping the temperature at <30°C. The mixture was then heated to 50°C for 2 hours. HPLC analysis [C-8 acetonitrile:water: phosphoric acid 30:70:0.1, 10 minute gradient to 80:20:0.1, 254 nm] shows no sulfide or sulfoxide remaining.

55 Sulfide RT = 9.9 min.  
Sulfoxide RT = 5.5 min.  
Sulfone RT = 7.7 min.

55 After cooling to room temperature 65 mL of saturated sodium bicarbonate was added over 5 minutes (gas evolution and the mixture was extracted with 55 ml of ethyl acetate. The aqueous layer was back extracted with 55 ml of ethyl acetate and the combined organics were washed with saturated sodium bicarbonate (2 x 65 ml) and 5% aqueous sodium chloride (50 mL). The organic extracts were concentrated in vacuo to a volume

of 20 ml. DMF (100 ml) was added and then concentrated in vacuo to 20 ml. The solution is used for the next step. HPLC typically shows 95% yield.

<sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>) δ 7.43 (d, J=1.8Hz, 1H), 7.20 (d, J=1.9Hz, 1H), 5.50 (dd, J=6.0, 8.7Hz, 1H), 4.12 (m, 2H), 3.96 (t, J=5.2Hz, 2H), 3.92 (s, 3H), 3.67-3.63 (m, 2H), 2.79-2.66 (m, 4H), 2.23-2.10 (m, 1H), 1.86 (sextet, J=7.2Hz, 2H), 1.03 (t, J=7.4Hz, 3H).

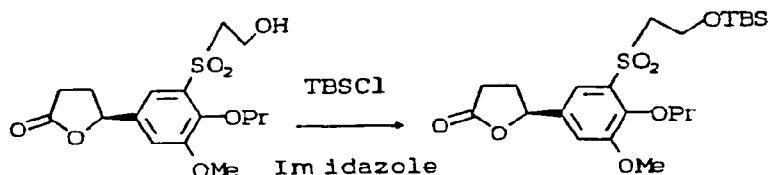
<sup>13</sup>CNMR (300 MHz, CDCl<sub>3</sub>) δ 176.3, 154.1, 146.8, 135.6, 133.1, 117.4, 114.7, 80.2, 76.5, 57.5, 56.5, 30.9, 29.1, 23.2, 10.3.

EXAMPLE 12

4-[3-Methoxy-4-n-propyloxy-5-(2'-t-butyl-dimethyl-siloxyethylsulfonyl)phenyl]-4-butyrolactone

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	<u>Materials</u>	<u>Amount</u>	<u>Moles</u>	<u>MW</u>
	4-[3-Methoxy-4-n-propyloxy-5-(2'-hydroxyethylsulfonyl)phenyl]-4-butyrolactone	3.0 g	8.38 mmol	358
	Imidazole	0.85 g	12.57 mmol	68
	t-butyldimethylsilyl-chloride	1.39 g	9.2 mmol	150.7
	DMF	3 ml		
	Ethyl Acetate	18 ml		
	Water	25 ml		
	5% Aqueous NaCl	50 ml		
	Toluene	50 ml		

5 Imidazole (0.85 g, 12.57 mmol) was added to a solution of sulfone (3.0 g, 8.38 mmol) in 6 ml of DMF (KF+278 µg/ml) at room temperature (25°C). A solution of t-butyldimethylsilylchloride (1.39 g, 9.2 mmol) in 3 ml of sieve dried DMF was added over 10 minutes keeping temperature  $\leq$ 30°C. The mixture was stirred at 25°C for 2 hours and the reaction followed by HPLC. HPLC assay [CH<sub>3</sub>CN:H<sub>2</sub>O:phosphoric acid 50:50:0.1 gradient to 80:20:0.1 over 8 minutes; C-8, 294 nm].

Alcohol RT = 3.0 min.

Silyl ether RT = 14.1 min.

10 Ethyl acetate (38 ml) was added and the mixture was washed with 25 ml water and then 2 x 25 ml with 5% aqueous sodium chloride. The organic extracts were concentrated *in vacuo* to a volume of 10 ml. Toluene (50 ml) was added and the solution was concentrated to a volume of 10 ml and checked by NMR for ethyl acetate (typically <5% EtOAc). HPLC assay typically shows 95% yield.

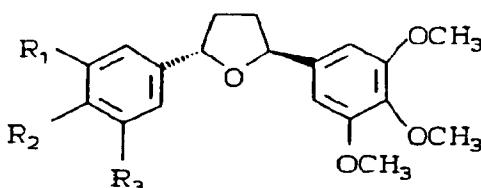
15 <sup>1</sup>H-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.38 (d, J=1.9Hz, 1H), 7.15 (d, J=1.9Hz, 1H), 5.46 (dd, J=5.8, 8.6Hz, 1H), 4.10 (m, 2H), 3.95 (t, J=6.1Hz, 2H), 3.89 (s, 3H), 3.72-3.57 (m, 2H), 2.67-2.61 (m, 3H), 2.20-2.10 (m, 1H), 1.86 (sextet, J=7.2Hz, 3H), 1.03 (t, J=7.4Hz, 3H), 0.73 (s, 9H), -0.092 (s, 3H), -0.097 (s, 3H). <sup>13</sup>C-NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  176.4, 154.0, 146.9, 135.1, 134.5, 117.2, 114.3, 80.4, 76.1, 57.6, 57.1, 56.4, 30.9, 29.1, 25.6, 23.2, 18.0, 10.3, -5.7.

### Claims

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1. A process of making a compound of Formula 2D:

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R<sup>1</sup> is S(O)<sub>n</sub>R<sub>a</sub> in which n is 0,1 or 2 and R<sub>a</sub> is selected from the group consisting of

- (a) C<sub>1-6</sub>alkyl,
- (b) protected hydroxy-C<sub>1-6</sub>alkyl, and
- (c) C<sub>1-6</sub>alkyl-carbonyl-C<sub>1-6</sub>alkyl;

R<sub>2</sub> is C<sub>1-6</sub>alkoxy; and

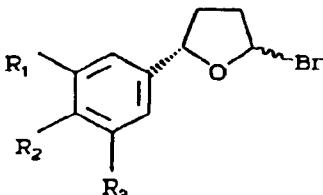
R<sub>3</sub> is selected from the group consisting of

- (a) C<sub>1-12</sub>alkoxy,
- (b) substituted C<sub>1-6</sub>alkoxy wherein the substituent is selected from the group consisting of protected hydroxy,
- (c) -O-C<sub>1-6</sub>alkyl-O-R<sup>10</sup>, wherein R<sup>10</sup> is

- (1) -PO<sub>2</sub>(OH)<sup>-</sup>M<sup>+</sup> wherein M<sup>+</sup> is a pharmaceutically acceptable cation,
- (2) -C(O)(CH<sub>2</sub>)<sub>2</sub>-CO<sub>2</sub><sup>-</sup>M<sup>+</sup>, or
- (3) -SO<sub>3</sub><sup>-</sup>M<sup>+</sup> comprising:

45 contacting of Compound C,

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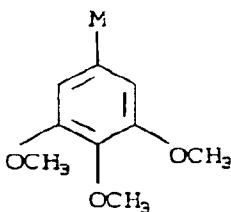
C

wherein the hydroxyl and amino groups of R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are protected, said Compound C contacting with

an organo metallic reagent species of the formula:

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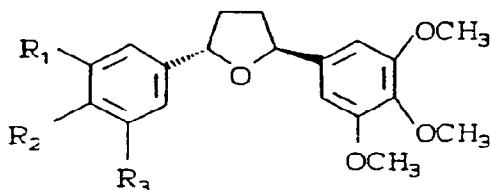


wherein M is selected from the group consisting of

- (a) copper,
- (b) magnesium,
- (c) aluminum,
- (d) lithium, and
- (e) zinc, in a etheral solvent to yield a compound of formula 2D:

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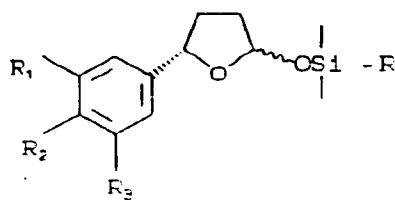
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wherein the etheral solvent is selected from the group consisting of, diethyl ether, di-n-butyl and diisopentyl ethers, anisole, tetrahydropyran, 4-methyl-1,3-dioxane, tetrahydrofurfuryl methyl ether, furan, tetrahydrofuran, dimethoxyethane, and diethoxymethane.

2. A process according to Claim 1 wherein the etheral solvent is tetrahydrofuran and M is copper.
3. A process according to Claim 1 wherein
  - R<sup>1</sup> is S(O)<sub>2</sub>R<sub>a</sub> wherein
  - R<sub>a</sub> is protected 2-hydroxyethyl, 2-oxopropyl or protected 2-hydroxypropyl,
  - R<sub>2</sub> is n-propoxy; and
  - R<sub>3</sub> is methoxy, 3-phosphopropoxy or protected 3-hydroxypropoxy.
4. A process according to Claim 1 where the compound of formula 2D is
  - (a) (-)-(2S,5S)-2-(3(2-hydroxyethylsulfonyl)-4-(n-propoxy)-5-methoxyphenyl)-5-(3,4,5-trimethoxyphenyl)tetrahydrofuran, a potent PAF antagonist;
  - (b) (-)-(2S,5S)-2-(3-(2-oxopropylsulfonyl)-4-(n-propoxy)-5-(3-phosphopropoxy) phenyl)-5-(3,4,5-trimethoxyphenyl)tetrahydrofuran;
  - (c) (-)-(2S,5S)-2-(3-(2-oxopropylsulfonyl)-4-(n-propoxy)-5-(3-hydroxypropoxy) phenyl)-5-(3,4,5-trimethoxyphenyl)tetrahydrofuran; and
  - (d) (-)-(2S,5S)-2-(3-(2-hydroxypropylsulfonyl)-4-(n-propoxy)-5-(3-hydroxypropoxy)phenyl)-5-(3,4,5-trimethoxyphenyl)tetrahydrofuran.
5. A process according to Claim 1 further comprising
  - contacting of Compound B

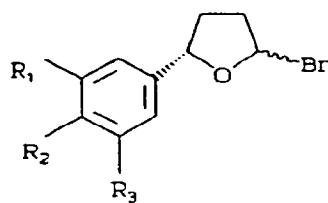
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10 wherein R is C<sub>1-6</sub> alkyl;

with a silyl bromide in a etheral or halo carbon solvent selected from the group consisting of mono or di halo C<sub>1-4</sub>alkyl, diethyl ether, di-n-butyl and diisopentyl ethers, anisole, tetrahydropyran, 4-methyl-1,3-dioxane, tetrahydrofurfuryl methyl ether, furan, tetrahydrofuran, methylene chloride, dimethoxyethane, diethoxymethane, to yield a glycosyl bromide Compound C;

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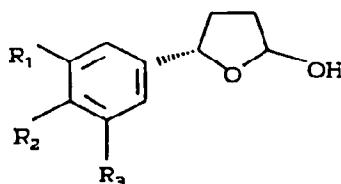
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wherein the hydroxyl groups of substituent R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are protected.

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6. A process according to claim 5 wherein the solvent is methylene chloride and the silyl bromide is trimethylsilyl bromide.
7. A compound selected from the group consisting of compound A'

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R<sup>1</sup> is S(O)<sub>n</sub>R<sub>a</sub> in which n is 0,1 or 2 and R<sub>a</sub> is selected from the group consisting of

- (a) C<sub>1-6</sub>alkyl,
- (b) protected hydroxy-C<sub>1-6</sub>alkyl, and
- (c) C<sub>1-6</sub>alkyl-carbonyl-C<sub>1-6</sub>alkyl;
- (d) 2-t-butyldimethylsiloxy ethyl;

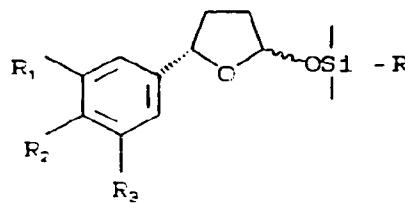
R<sub>2</sub> is C<sub>1-12</sub>alkoxy; and

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R<sub>3</sub> is selected from the group consisting of

- (a) C<sub>1-6</sub>alkoxy,
- (b) substituted C<sub>1-6</sub>alkoxy wherein the substituent is selected from the group consisting of protected hydroxy;
- (c) -0-C<sub>1-6</sub>alkyl-0-R<sup>10</sup>, wherein R<sup>10</sup> is
  - (1) -PO<sub>2</sub>(OH)<sup>-</sup> M<sup>+</sup> wherein M<sup>+</sup> is a pharmaceutically acceptable cation,
  - (2) -C(0)(CH<sub>2</sub>)<sub>2</sub>-CO<sub>2</sub><sup>-</sup>M<sup>+</sup>, or
  - (3) -SO<sub>3</sub><sup>-</sup>M<sup>+</sup>; compound B

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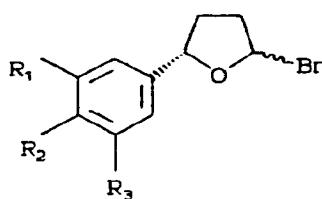
B

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wherein R is C<sub>1-6</sub> alkyl; and,  
compound C

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C

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with the proviso that the hydroxyl groups of compound C are protected.

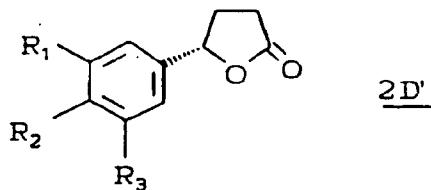
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8. A Compound of formula B or C according to Claim 7 wherein  
R<sup>1</sup> is S(0)<sub>2</sub>R<sub>a</sub> wherein  
R<sub>a</sub> is protected 2-hydroxyethyl, 2-oxopropyl, 2-t-butyldimethyl siloxyethyl or protected 2-hydroxypropyl,  
R<sub>2</sub> is n-propoxy; and  
R<sub>3</sub> is methoxy, 3-phosphopropoxy or protected 3-hydroxypropoxy.

9. A process of making a Compound of Formula 2D'

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2D'

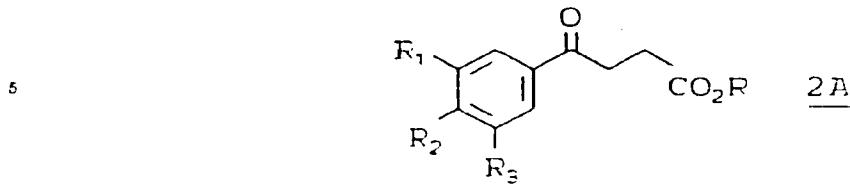
wherein

R<sub>1</sub> is iodide or bromide,  
R<sub>2</sub> is C<sub>1-12</sub>alkoxy; and

R<sub>3</sub> is selected from the group consisting of

- (a) C<sub>1-6</sub>alkoxy,
- (b) substituted C<sub>1-6</sub>alkoxy wherein the substituent is selected from the group consisting of protected hydroxy,
- (c) -O-C<sub>1-6</sub>alkyl-O-R<sup>10</sup>, wherein R<sup>10</sup> is
  - (1) -PO<sub>2</sub>(OH)<sup>-</sup> M<sup>+</sup> wherein M<sup>+</sup> is a pharmaceutically acceptable cation,
  - (2) -C(O)(CH<sub>2</sub>)<sub>2</sub>-CO<sub>2</sub><sup>-</sup>M<sup>+</sup>, or
  - (3) -SO<sub>3</sub><sup>-</sup>M<sup>+</sup> comprising contacting a compound of Formula 2A

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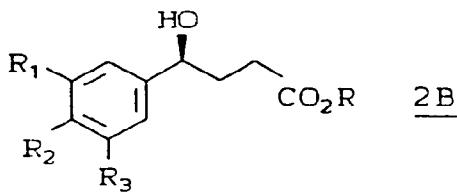


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in an ethereal solvent selected from the group consisting of diethyl ether, di-n-butyl and diisopentyl ethers, anisole, cyclic ethers such as tetrahydropyran, 4-methyl-1,3-dioxane, dihydropyran, tetrahydrofuryl methyl ether, ethyl ether, furan, 2-ethoxytetrahydrofuran with  $\beta$ -chlorodiisopinocamphenyl borane to yield a compound of formula 2B

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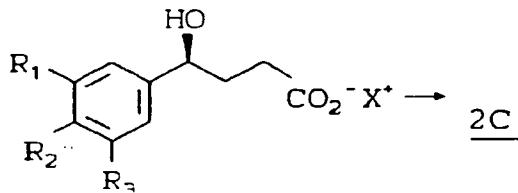


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contacting compound 2B in a medium containing alcohol in an ethereal solvent selected from the group consisting of diethyl ether, di-n-butyl and diisopentyl ethers, anisole, cyclic ethers such as tetrahydropyran, 4-methyl-1,3-dioxane, dihydropyran, tetrahydrofuryl methyl ether, ethyl ether, furan, 2-ethoxytetrahydrofuran with an alkali metal hydroxide to yield a compound of formula 2C,

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wherein X is an alkali metal selected from the group consisting of sodium, potassium and lithium.

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10. A process according to Claim 9 wherein

R<sub>1</sub> is iodide or bromide,

R<sub>2</sub> is C<sub>1-12</sub>alkoxy; and

R<sub>3</sub> is selected from the group consisting of

(a) C<sub>1-6</sub>alkoxy,

(b) substituted C<sub>1-6</sub>alkoxy wherein the substituent is selected from the group consisting of hydroxy and amino,

(c) -O-C<sub>1-6</sub>alkyl-O-R<sup>10</sup>, wherein R<sup>10</sup> is

(1) -PO<sub>2</sub>(OH)<sup>-</sup> M<sup>+</sup> wherein M<sup>+</sup> is a pharmaceutically acceptable cation,

(2) -C(O)(CH<sub>2</sub>)<sub>2</sub>-CO<sub>2</sub><sup>-</sup> M<sup>+</sup>, or

(3) -SO<sub>3</sub><sup>-</sup> M<sup>+</sup>.

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11. A process according to Claim 10 wherein

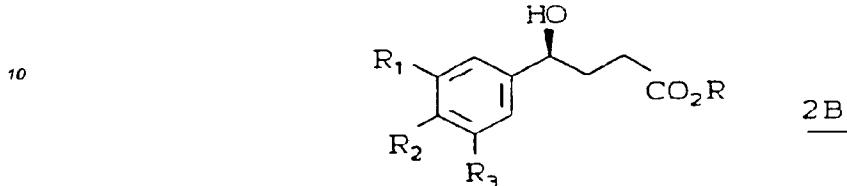
R<sub>2</sub> is n-propoxy; and

R<sub>3</sub> is methoxy, 3-phosphopropoxy or 3-hydroxypropoxy.

12. A process according to claim 9 wherein Compound D is selected from the group consisting of

(a) 4-(3-methoxy-4-n-propyl-5-iodophenyl)-4-butyrolactone,  
 (b) 4-(3-phosphopropoxy-4-n-propyl-5-iodophenyl)-4-butyrolactone, or  
 (c) 4-(3-hydroxypropoxy-4-n-propyl-5-iodophenyl)-4-butyrolactone.

## 5 13. A compound of formula 2B



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wherein

R is CO<sub>2</sub>Et, CO<sub>2</sub>Me, CO<sub>2</sub>CH<sub>2</sub>Ph, CO<sub>2</sub>CH<sub>2</sub>CHCH<sub>2</sub>, CO<sub>2</sub>Ph, CO<sub>2</sub>-I-C<sub>4</sub>H<sub>9</sub> or CN, and  
 R<sub>1</sub> is iodide or bromide;

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R<sub>2</sub> is C<sub>1-12</sub>alkoxy; andR<sub>3</sub> is selected from the group consisting of

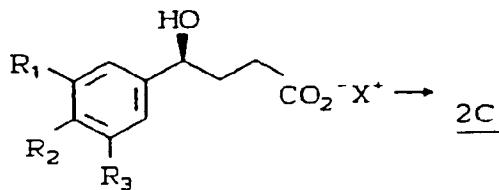
(a) C<sub>1-6</sub>alkoxy,  
 (b) substituted C<sub>1-6</sub>alkoxy wherein the substituent is selected from the group consisting of hydroxy and amino,  
 (c) -O-C<sub>1-6</sub>alkyl-0-R<sup>10</sup>, wherein R<sup>10</sup> is  
 (1) -PO<sub>2</sub>(OH)<sup>-</sup>M<sup>+</sup> wherein M<sup>+</sup> is a pharmaceutically acceptable cation,  
 (2) -C(O)(CH<sub>2</sub>)<sub>2</sub>-CO<sub>2</sub><sup>-</sup>M<sup>+</sup>, or  
 (3) -SO<sub>3</sub><sup>-</sup>M<sup>+</sup>.

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## 14. A compound of Formula 2C

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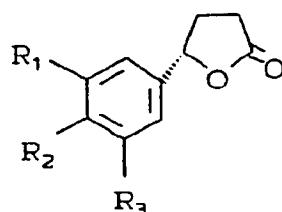
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wherein X is an alkali metal selected from the group consisting of sodium, potassium and lithium, and

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## 15. A compound according to Claim 14 wherein

R<sub>1</sub> is iodide or bromide,R<sub>2</sub> is C<sub>1-12</sub>alkoxy; andR<sub>3</sub> is selected from the group consisting of

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(a) C<sub>1-6</sub>alkoxy,  
 (b) substituted C<sub>1-6</sub>alkoxy wherein the substituent is selected from the group consisting of hydroxy and amino,

(c) -O-C<sub>1-6</sub>alkyl-0-R<sup>10</sup>, wherein R<sup>10</sup> is

- (1) -PO<sub>2</sub>(OH)<sup>-</sup> M<sup>+</sup> wherein M<sup>+</sup> is a pharmaceutically acceptable cation,
- (2) -C(O)(CH<sub>2</sub>)<sub>2</sub>-CO<sub>2</sub><sup>-</sup>M<sup>+</sup>, or
- (3) -SO<sub>3</sub><sup>-</sup>M<sup>+</sup>.

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16. A compound according to Claim 15 wherein

R<sub>2</sub> is n-propoxy; and

R<sub>3</sub> is methoxy, 3-phosphopropoxy or 3-hydroxypropoxy.

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## EUROPEAN SEARCH REPORT

DOCUMENTS CONSIDERED TO BE RELEVANT			EP 91305816.0						
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CL.5)						
P, X	EP - A1 - 0 402 150 (MERCH) * Claim 1; page 11; formula 19 * --	1, 7, 8	C 07 D 307/12 C 07 C 69/732 C 07 C 255/15						
P, X	CHEMICAL ABSTRACTS, vol. 113, no. 1, July 2, 1990, Columbus, Ohio, USA GUTMANN, ARIE L. et al. "Lipase-catalyzed preparation of optically active gamma- -butyrolactones in organic solvents" page 589, column 1, abstract-no. 6 066z & J. Org. Chem. 1990, 55(11) .. 3546-52 --	13, 14							
P, X	CHEMICAL ABSTRACTS, vol. 114, no. 17, April 29, 1991, Columbus, Ohio, USA THOMPSON, A.S. et al. "Conversion of a silylated hemiacetal into an alpha- -bromoether using trimethyl- silyl bromide. Synthesis of platelet activating factor antagonist L-659.989." page 745, column 1, abstract-no. 163 901f & Tetrahedron Lett. 1990, 31(48), 6953-6 (Eng). --	1, 7, 8							
D, X	CHEMICAL ABSTRACTS, vol. 108, no. 9, February 29, 1988, Columbus, Ohio, USA COREY, E.J. et al. "A stable and easily prepared catalyst for the enantioselective re- duction of ketones. Appli-	7, 8							
<p>The present search report has been drawn up for all claims</p> <table border="1"> <tr> <td>Place of search VIENNA</td> <td>Date of completion of the search 03-09-1991</td> <td>Examiner HAMMER</td> </tr> <tr> <td colspan="2">CATEGORY OF CITED DOCUMENTS</td> <td>           T : theory or principle underlying the invention            E : earlier patent document, but published on, or            after the filing date            D : document cited in the application            L : document cited for other reasons            A : technological background            O : non-written disclosure            P : intermediate document            &amp; : member of the same patent family, corresponding            document         </td> </tr> </table>				Place of search VIENNA	Date of completion of the search 03-09-1991	Examiner HAMMER	CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons A : technological background O : non-written disclosure P : intermediate document & : member of the same patent family, corresponding document
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CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons A : technological background O : non-written disclosure P : intermediate document & : member of the same patent family, corresponding document							



-2-

EP 91305816.0

DOCUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int. CL.5)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
	cations to multistep syntheses." page 644, column 1, abstract-no. 75 026v & J. Am. Chem. Soc. 1987, 109(25), 7925-6 (Eng).		
A	<u>GB - A - 2 197 650</u> (MERCH) * Claims 1-10 *	1	
A	<u>EP - A2 - 0 199 324</u> (MERCH) * Column 9,10; claims 3,4 *	1	
1	CHEMICAL ABSTRACTS, vol. 111, no. 7, August 14, 1989, Columbus, Ohio, USA Ponpipom, M.M. et al. "Enantioselective cyclization of chiral butane-1,4-diols to chiral tetrahydrofurans: synthesis of chiral trans- -2-(3-methoxy-5-(methyl- sulfonyl)-4-propoxypyphenyl)- -5-(3,4,5-trimethoxyphenyl)- tetrahydrofuran (L-659,989), a potent PAF-receptor antagonist." Page 731, column 1, abstract- no. 57 433h & Tetrahedron Lett. 1988, 29(48, 6211-14 (Eng).	A	
A	<u>EP - A2 - 0 322 033</u> (MERCH) * Pages 8-13; claims 9,10 *		
The present search report has been drawn up for all claims			
Place of search	Date of completion of the search	Examiner	
VIENNA	03-09-1991	HAMMER	
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant H taken alone	T : theory or principle underlying the invention		
Y : particularly relevant H combined with another document of the same category	E : earlier patent document, but published on, or after the filing date		
A : technological background	D : document cited in the application		
O : non-written disclosure	L : document cited for other reasons		
P : intermediate document	& : member of the same patent family, corresponding document		